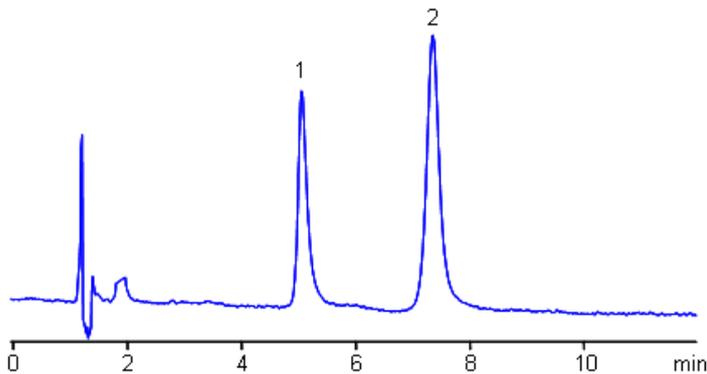


## Separation of Benzoic and Acetylbenzoic acid in Hydrogen-Bonding Mode

**Column:** SHARC 1  
**Size:** 3.2 x 100 mm  
**Mobile phase:** MeCN/MeOH 95/5 with 10 mM Ammonia  
**Flow:** 0.5 mL/min  
**Detection:** UV 250 nm

1. Benzoic acid
2. 4-Acetylbenzoic acid



A general approach for analysis of various organic acids with MS detection in negative mode was developed using hydrogen-bonding stationary phase – SHARC1. A highly sensitive method allows to analyze traces of organic acids in various matrices using ACN/MeOH/ammonia mobile phase. In hydrogen-bonding chromatography acetonitrile is a weaker solvent and alcohol is a stronger solvent. Various gradient and isocratic conditions can be used

The SHARC™ family of innovative columns represents the first commercially available columns primarily utilizing separation based on hydrogen bonding. SHARC stands for Specific Hydrogen-bond Adsorption Resolution Column. Hydrogen bonding involves an interaction or attraction between a bound hydrogen atom and molecules containing electronegative atoms, such as oxygen, nitrogen, and fluorine.

### Method Parameters

<b>Column</b>	Sharc 1, 3.2x100 mm, 5 µm, 100 Å
<b>Mobile Phase</b>	MeCN/MeOH – 95/5%
<b>Buffer</b>	Ammonia 10mM
<b>Flow Rate</b>	005 mL/min
<b>Detection</b>	UV, 250 nm

Quelle: <https://sielc.com/Separation%20of%20Benzoic%20and%20Acetylbenzoic%20acid%20in%20Application-Hydrogen-Bonding-Mode>